

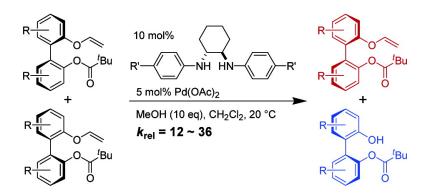
Communication

Kinetic Resolution of Axially Chiral 2,2'-Dihydroxy-1,1'-biaryls by Palladium-Catalyzed Alcoholysis

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J. Am. Chem. Soc., 2005, 127 (30), 10474-10475• DOI: 10.1021/ja051750h • Publication Date (Web): 08 July 2005

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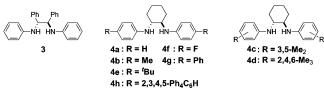
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Axially chiral biaryls are important modules in synthetic,^{1a} pharmaceutical,^{1b} material,^{1c} and supramolecular^{1d} chemistry. Especially, optically active 2,2'-dihydroxy-1,1'-biaryls have received primary attention because they have been utilized not only as ligands but also as precursors for phosphine ligands in catalytic asymmetric reactions.^{1a} The success of complexes of 2,2'-disubstituted 1,1'-binaphthyls, in particular, BINOL and BINAP, in giving high enantioselectivities in numerous catalytic reactions has encouraged the synthesis of several related ligands.

The optically active 2,2'-dihydroxy-1,1'-biaryls have been provided by conventional optical resolution or chiral pool methods using stoichiometric chiral sources² because they cannot be constructed by the typical catalytic asymmetric reactions, including hydrogenation and aldol reaction. The enantioselective oxidative coupling of 2-naphthols is the only way to access them so far. However, this method is effective only for the building of optically active BINOL and its analogues.³ The efficient synthesis of optically active 1,1'-bi-2-phenols has not been attained due to the low reactivity of phenols. Thus, the development of a novel methodology for these classes of compounds is an interesting target in asymmetric catalysis.



We recently reported a catalytic hydrolysis of alkenyl ethers and esters and their application to hydrolytic kinetic resolution⁴ of some racemic vinyl ethers.⁵ Although various metal complexes, such as Pd^{II}, Pt^{II}, Hg^{II}, Cu^{II}, Co^{III}, Ru^{II}, and Sc^{III}, showed activity in the hydrolysis of vinyl ethers, only (salen)Co complexes catalyzed the asymmetric hydrolysis, giving moderate selectivity ($k_{rel} = 10$). Herein, we report a Pd-catalyzed kinetic resolution of vinyl ethers of axially chiral 2,2'-dihydroxy-1,1'-biaryls with high selectivity and generality. This is the first example for the efficient preparation of optically active 1,1'-bi-2-phenols by a catalytic system.

Our attempts started with the hydrolysis of BINOL vinyl ethers using (salen)Co and chiral ligands–Pd systems. The optimization study revealed that chiral secondary diamines 4/Pd-catalyzed methanolysis⁶ rather than hydrolysis gave favorable results. The kinetic resolution of racemic **1a** by Pd(OAc)₂ complexes was performed under the methanolysis condition (Table 1). A diamine ligand **4a** derived from (*R*,*R*)-1,2-cyclohexanediamine showed higher selectivity ($k_{rel} = 16.5$, entry 4) than that of another diamine ligand **3** ($k_{rel} = 7.2$, entry 3) and typical chiral ligands, such as BINAP ($k_{rel} = 1.3$, entry 1) and sparteine ($k_{rel} = 1.4$, entry 2). The selectivity was largely influenced by the position of substituents on the phenyl ring of **4**, the ligand **4b** having a methyl group at the **Table 1.** Kinetic Resolution of rac-**1a** Catalyzed by $Pd(OAc)_2$ Complexes with Various Kinds of (R,R)-Diamine^a

	10 mol	5 mol% Pd(OAc) ₂ 10 mol% ligand 10 eq MeOH CH ₂ Cl ₂ , 20 °C			+ OH OH O		
rac-	-1a		1a	2a			
		t	conv	ee of 1a	ee of 2a		
entry	ligand	(h)	(%) ^b	(%) ^c	(%) ^c	<i>k</i> _{rel} ^d	
1	(S)-BINAP	137	22	3	15	1.3	
2	(-)-sparteine	187	9	2	20	1.4	
3	3	18	43	50	64	7.2	
4	4a	62	43	61	81	16.5	
5	4b	109	40	54	81	16.0	
6	4c	107	19	16	68	5.8	
7	4d	46	41	2	3	1.1	
8	4e	111	11	11	88	14.7	
9	4f	42	41	56	79	13.8	
10	4g	61	35	45	84	18.3	
11	4h	61	58	96	69	20.3	

^{*a*} The reaction was carried out with 1 M solution of **1a**, 5 mol % of Pd(OAc)₂, 10 mol % of ligand, and 10 equiv of MeOH in CH₂Cl₂ at 20 °C. ^{*b*} Calculated from isolated yields of **1a** and **2a**. ^{*c*} Determined by HPLC. ^{*d*} From ref 10.

para-position exhibited higher k_{rel} value ($k_{rel} = 16.0$, entry 5) than *meta*- (**4c**, $k_{rel} = 5.8$, entry 6) and *ortho*- (**4d**, $k_{rel} = 1.1$, entry 7) positions. The steric and electronic effect of the *para*-substituent was investigated (H, Me, 'Bu, F, and Ph, entries 4, 5, 8, 9, and 10). The ligand **4g** bearing a phenyl group gave slightly higher selectivity ($k_{rel} = 18.3$, entry 10) than did **4a**, although other ligands showed lower selectivity. On the basis of this result, we introduced a 2,3,4,5-tetraphenylphenyl group,⁷ which exhibited unique steric effect in a Pd/pyridines-catalyzed aerobic oxidation of alcohols.⁸ Interestingly, **4h** afforded the best result with regard to selectivity ($k_{rel} = 20.3$) and reactivity (entry 11), while the typical bulky substituent (*t*-Bu) suffered from low reactivity (entry 8).

To explore the effect of bulkiness of the acyl group on BINOL, we examined a series of 2-acyloxy-2'-vinyloxy-1,1'-binaphthyls with Pd(OAc)₂-**4h** as catalyst (Table 2). The nonacylated compound **1b** showed almost no selectivity ($k_{rel} = 1.1$), although the reactivity was extremely high (entry 1). The k_{rel} value was increased by larger substituent as follows: acetyl (**1c**, $k_{rel} = 6.1$, entry 2), 1-heptanoyl (**1d**, $k_{rel} = 14.3$, entry 3), pivaloyl (**1a**, $k_{rel} = 20.3$, entry 4), and 1-adamantanoyl (**1e**, $k_{rel} = 28.7$, entry 5).

Then, we investigated the reaction with various kinds of racemic 2-pivaloyloxy-2'-vinyloxy-1,1'-binaphthyls and 1,1'-biphenyls using Pd(OAc)₂-**4h** as catalyst (Table 3). The reaction system was applicable to all examined substrates **1a** and **1f**-**m**, giving moderate to high selectivity ($k_{rel} = 12.1-35.8$). For example, the kinetic resolution of **1i** proceeded with k_{rel} values of about 30. The

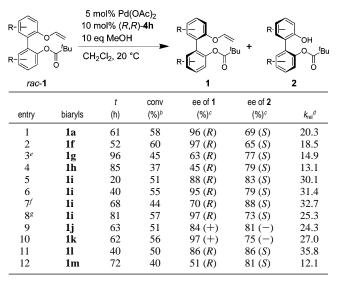
Table 2. Kinetic Resolution of

2-Acyloxy-2'-vinyloxy-1,1'-binaphthyl 1 Catalyzed by Pd(OAc)2-4h Complex^a

(5 mol% Pd(OAc) ₂ 10 mol% (<i>R,R</i>)- 4h 10 eq MeOH		\mathbf{i}	`0´\ +	\square	ОН
OR		CH ₂ Cl ₂ , 20 °C	OR .			OR	
rac-1			ب ١			~	2
			t	conv	ee of 1	ee of 2	
entry	BINOL	R	(h)	(%) ^b	(%) ^c	(%) ^c	$k_{\rm rel}{}^d$
1	1b	Н	3	43	4	6	1.1
2	1c	COCH ₃	24	49	54	57	6.1
3	1d	$CO(n-C_6H_{13})$	63	44	61	77	14.3
4	1a	CO(t-Bu)	61	58	96	69	20.3
5	1e	CO(1-adamantyl)	42	56	96	77	28.7

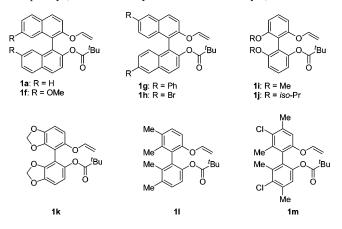
^a The reaction was carried out with 1 M solution of 1, 5 mol % of Pd(OAc)₂, 10 mol % of (R,R)-4h, and 10 equiv of MeOH in CH₂Cl₂ at 20 °C. ^b Calculated from isolated yields of 1 and 2. ^c Determined by HPLC. ^d From ref 10.

Table 3. Kinetic Resolution of Binaphthyls and Biphenyls Catalyzed by Pd(OAc)₂-4h Complex^a



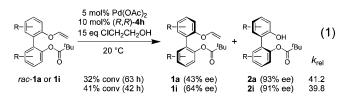
^a The reaction was carried out with 1 M solution of 1, 5 mol % of Pd(OAc)₂, 10 mol % of (R,R)-4h, and 10 equiv of MeOH in CH₂Cl₂ at 20 °C. ^b Calculated from isolated yields of 1 and 2. ^c Determined by HPLC. ^d From ref 10. ^e At 0.7 M condition. ^f With 1 mol % of Pd(OAc)₂ and 2 mol % of (R,R)-4h used. ^g (R,R)-4a was used as ligand.

enantiomeric excess of 1i increased from 88 to 95% (20 h entry 5, and 40 h entry 6) with the increase in conversion. Even in the low catalyst loading condition (1 mol %), the selectivity was retained completely ($k_{rel} = 32.7$, entry 7, cf. $k_{rel} = 30.1$, entry 5). In addition,



the deference in reactivity and selectivity between ligand 4h and 4a was again observed (4a, $k_{rel} = 25.3$, entry 8, cf. entry 5). The highest $k_{\rm rel}$ value of 35.8 was observed with **11** (entry 11).

Although methanol was the best reagent in the present alcoholysis reaction in terms of reactivity, the $k_{\rm rel}$ value was improved to 41.2 (1a) and 39.8 (1i) by the use of 2-chloroethanol as reagent (eq 1).



The reaction appeared to take place in a similar manner with Pd-catalyzed transfer vinylation from vinyl ethers to alcohols,9 although our work is the first example of its asymmetric version. Actually, 1-dodecyl vinyl ether was isolated using 1-dodecanol as reagent, though the k_{rel} value was decreased to 3.6. We expect that the reaction is irreversible. In fact, the vinylation of 2a by ethyl vinyl ether using Pd(OAc)₂-4h did not proceed.9

In conclusion, we have achieved palladium-catalyzed kinetic resolution of various kinds of 2,2'-dihydroxy-1,1'-biaryls by alcoholysis reaction of their vinyl ethers. The reaction was applicable to 1,1'-bi-2-phenols as well as 1,1'-bi-2-naphthols with high selectivity.

Acknowledgment. This work was supported by Grant-in-Aid for Scientific Research on Priority Areas (No. 16033204, Reaction Control of Dynamic Complexes) from Ministry of Education, Culture, Sports, Science and Technology, Japan.

Supporting Information Available: Experimental procedures and full characterization of new compounds (PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

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